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Nitric Acid Uptake on	Subtropical Cirrus	Cloud Particles
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Abstract

2 The redistribution of HNO₃ via uptake and sedimentation by cirrus cloud particles is considered an important term in the upper tropospheric budget of reactive nitrogen. 4 Numerous cirrus cloud encounters by the NASA WB-57F high-altitude research aircraft during CRYSTAL-FACE were accompanied by the observation of condensed-phase 6 HNO₃ with the NOAA chemical ionization mass spectrometer. The instrument measures HNO₃ with two independent channels of detection connected to separate forward- and 8 downward-facing inlets that allow a determination of the amount of HNO₃ condensed on ice particles. Subtropical cirrus clouds, as indicated by the presence of ice particles, were 10 observed coincident with condensed-phase HNO3 at temperatures of 197 K - 224 K and pressures of 122 hPa - 224 hPa. Maximum levels of condensed-phase HNO₃ approached 12 the gas-phase equivalent of 0.8 ppbv. Ice particle surface coverages as high as $1.4 \cdot 10^{14}$ molecules·cm⁻² were observed. A dissociative Langmuir adsorption model, when using 14 an empirically derived HNO₃ adsorption enthalpy of -11.0 kcal·mol⁻¹, effectively describes the observed molecular coverages to within a factor of 5. The percentage of 16 total HNO₃ in the condensed phase ranged from near zero to 100% in the observed cirrus clouds. With volume-weighted mean particle diameters up to 700 μ m and particle fall 18 velocities up to 10 m·s⁻¹, some observed clouds have significant potential to redistribute HNO₃ in the upper troposphere.

1. Introduction

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2 Cirrus clouds are ubiquitous throughout the upper troposphere (UT) and can cover as much as 40% of the Earth's surface [Liao et al., 1995, Jin et al., 1996, Wang et al., 1996, 4 Wylie and Menzel, 1999]. Composed of ice crystals [Lynch, 2002], cirrus clouds are known to play a complex and significant role in the global radiation budget [Liou, 1986]. Cirrus clouds can be formed in situ in the UT [Kärcher, 2002], as a result of synoptic 6 weather disturbances, or in the anvil outflow at the top of cumulonimbus clouds [Sassen, 8 2002]. Tropical cirrus clouds around the peak convective detrainment level are formed primarily via the latter mechanism, and can reach altitudes of up to 18 km when produced 10 in deep convective systems. The broad lateral and vertical extent of anvil cirrus clouds produced in the tropics is expected to exert a greater influence on the Earth's climate 12 system than midlatitude cirrus [Heymsfield and McFarquhar, 2002]. Due to the high altitudes and often-remote locations of tropical cirrus, however, comprehensive in situ 14 measurements of these clouds have been limited.

Model simulations by Lawrence and Crutzen [1998] suggest that the uptake and gravitational redistribution of nitric acid (HNO₃) by cirrus cloud particles may represent a significant sink of HNO₃ in the UT. Nitric acid serves as a primary reservoir species for nitrogen oxides (NO_x) [Neuman et al., 2001], which are directly involved in the photochemical production of tropospheric ozone [Jaeglé et al., 1998]. Model studies of cirrus-processed air have demonstrated that the sedimentary removal of HNO₃ from the UT can effect strong local reductions in NO_x, with the consequence of significant reductions in the net ozone production rate [Meier and Hendricks, 2002]. Since ozone is known to be an effective greenhouse gas in the troposphere [Albritton et al., 2001], particularly near the tropopause [Lacis et al., 1990], understanding the uptake and redistribution of HNO₃ by cirrus cloud particles may be important in assessing the contribution of cirrus clouds to the radiative forcing of climate change.

A number of laboratory studies have investigated the uptake of HNO₃ on ice surfaces

at temperatures typical of the UT. Experiments performed by Zondlo et al. [1997] on vapor-deposited ice films at 211 K resulted in observed HNO3 surface coverages of 2 1.5·10¹⁵ molecules·cm⁻². A series of similar experiments reported by *Hudson et al.* [2002] at an HNO₃ pressure ($P(HNO_3)$) of 1.1·10⁻⁶ hPa indicated a negative temperature 4 dependence to the observed uptake, with coverages of 1.1·10¹⁴ to 5.9·10¹³ molecules·cm⁻ ² over a temperature range of 214-220 K. Uptake studies performed on ice films by 6 Abbatt [1997] yielded coverages of up to 2.9·10¹⁴ molecules·cm⁻² at temperatures as low as 208 K, with no apparent dependency on $P(\text{HNO}_3)$ values over the range 1.7·10⁻⁷ to 8 4.1·10⁻⁶ hPa. Hynes et al. [2002] reported comparable coverages using a similar 10 technique, although they observed coverages increasing by factor of 2 over a nearly 10fold increase in P(HNO₃), from 5.0·10⁻⁷ - 3.0·10⁻⁶ hPa. A laboratory study of HNO₃ uptake on nebulized half-micron diameter ice particles at 230 K yielded coverages similar 12 to those observed on the ice films (1.2·10¹⁴ molecules·cm⁻²), although these experiments were performed at the relatively high P(HNO₃) of 7·10⁻⁶ hPa [Arora et al., 1999]. There 14 has not yet been an extensive laboratory study of HNO3 uptake on ice surfaces performed at $P(HNO_3)$ values typical of the UT (< 2.0·10⁻⁷ hPa). 16 Prior field studies of HNO₃ uptake on cirrus cloud particles have been made at mid and high latitudes. Measurements of total reactive nitrogen ($NO_y = NO + NO_2 + 2N_2O_5$ 18 + HNO₃ +...) in a mountain wave cloud over the continental United States reported by 20 Weinheimer et al. [1998] indicated that levels of condensed NO_v in the cloud approached 20% of total NO_v. Surface coverages on the wave cloud ice particles were calculated to be as high as 2.5·10¹³ molecules·cm⁻² [Hudson et al., 2002]. Measurements of 22 condensed-phase NO_v in cirrus layers in the Arctic UT by Kondo et al. [2003] yielded HNO₃ coverages as high as 1.6·10¹⁴ molecules·cm⁻² at temperatures of approximately 24 200 K, with coverages decreasing at warmer temperatures. Meilinger et al. [1999] 26 conducted similar measurements in Arctic cirrus clouds at 196 K and reported coverages

of only 1·10¹³ molecules·cm⁻². An extensive dataset of measurements made at

midlatitudes revealed median levels of condensed NO_y (assumed to be HNO₃) in the northern hemisphere of $3.6 \cdot 10^{12}$ molecules·cm⁻² to be greater than twice that observed in the southern hemisphere [Ziereis et al., 2003]

We report here an extensive dataset of in situ measurements, including gas- and condensed-phase HNO₃ and ice particle surface area density (SAD), obtained in subtropical in situ and anvil cirrus clouds. These measurements were conducted onboard the NASA WB-57F high-altitude research aircraft as part of the Cirrus Regional Study of Tropical Anvils and Cirrus Layers Florida Area Cirrus Experiment (CRYSTAL-FACE) mission. The data are used here to assess the uptake of HNO₃ by subtropical cirrus cloud particles and explore the partitioning of HNO₃ between the gas and ice particle phases in cirrus clouds.

2. Instrumentation

This study utilizes data from a number of in situ instruments onboard the NASA WB-57F aircraft. Gas-phase and condensed-phase HNO₃ measurements made by chemical ionization mass spectrometry are described in more detail below. Particle size distribution and number density measurements made by the Cloud, Aerosol and Precipitation Spectrometer (CAPS) were used to derive SAD and volume-weighted mean diameter (VMD) for ice particles in the size range between 0.35-1550 μm [Baumgardner et al., 2001]. Note that all particle sizes cited herein refer to particle diameter, and not radius. A second, independent measurement of SAD was provided by the Cloud Integrating Nephelometer (CIN) [Gerber et al., 2000]. Ice water content (IWC) and water (H₂O) vapor were measured by the Harvard University Lyman-α hygrometer [Weinstock et al., 1994, Weinstock et al., 2003]. Nitric oxide (NO) and total reactive nitrogen (NO_y) were measured by catalytic reduction and chemiluminescence [Weinheimer et al., 2003]. Ambient temperature and pressure, and WB-57F true air speed were measured by the Meteorological Measurement System (MMS) [Scott et al., 1990]. The precision and accuracy of these measurements are summarized in Table 1.

Tropopause height was measured by the microwave temperature profiler (MTP)

[Denning et al., 1989].

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2.1. HNO₃ Measurements

HNO₃ was measured using the NOAA chemical ionization mass spectrometer 4 (CIMS) located in the third pallet position of the NASA WB-57F aircraft. This 6 instrument measures HNO₃ with an accuracy of ±20% and precision of 30 pptv (10, 10-s averages), and has been described in detail elsewhere [Neuman et al., 2000]. Prior to 8 CRYSTAL-FACE, the NOAA CIMS was modified by the addition of a second independent channel for the measurement of HNO₃ and the relocation of the original 10 sample inlet on the CIMS inlet pylon (Fig. 1). The two CIMS channels are designed to provide identical measurements of gas-phase HNO₃. Due to differences in the particle 12 sampling efficiencies of the two inlets, however, the two channels have different sensitivity to condensed-phase HNO3. When sampling in cirrus clouds, the forwardfacing front inlet samples both gas-phase HNO3 and any HNO3 condensed on the cirrus 14 particles. The downward-facing bottom inlet samples primarily gas-phase HNO3 because 16 the plane of the sampling orifice is parallel to the direction of flow over the inlet, which is set by the flow straightener. Semi-empirical calculations indicate that approximately 18 50% of 0.1 µm particles and greater than 90% of 1 µm and larger particles are inertially stripped from the air sampled by the bottom inlet [Vincent et al., 1986]. Calculations 20 further suggest that ice particles greater than 10 μ m in diameter (typical of cirrus clouds observed during CRYSTAL-FACE) are almost entirely removed from the sampled air. 22 Thus, for most of the cirrus clouds sampled, the HNO₃ from the bottom inlet is taken to be a measure of the gas-phase HNO₃ abundance.

The conclusion that ice particles greater than approximately 1 μ m are inertially separated from air sampled by the bottom inlet is further supported by measurements made in the contrail of the WB-57F during CRYSTAL-FACE. The contrail contained ice particles with high number densities (100-200 cm⁻³) and volume-weighted mean

diameters on the order of 2 μ m [Gao et al., 2003]. HNO₃ was present in the contrail because of mixing between the exhaust gases and ambient air containing approximately 0.4 ppbv HNO₃. Measurements in the contrail as soon as 4 minutes after formation indicated a difference between the front and bottom CIMS channels. The minimum signal from the bottom channel was near zero inside the contrail. This low HNO₃ signal is consistent with the removal of gas-phase HNO₃ by uptake onto the 2- μ m ice particles in the contrail, and the inertial stripping of these particles from the bottom inlet sample flow. As expected, a simultaneous increase in HNO₃ above ambient values occurred in the front CIMS channel, which does not discriminate against 2 μ m particles. If the bottom CIMS inlet sampled 2 μ m particles with any significant efficiency, HNO₃ observed in the bottom channel during the contrail intercept would not be significantly lower than the ambient values immediately outside the contrail.

The front CIMS inlet samples subisokinetically, meaning the sample air velocity inside the inlet (U) is less than the WB-57F true air speed (U_0) of 140-200 m·s⁻¹ at sampling altitudes. As a result, cirrus cloud particle number densities in the sampled air stream are inertially enhanced relative to those in the ambient air. A computational fluid dynamics program (Fluent Inc., New Hampshire) was used to estimate particle enhancement factors (EF) in the front inlet by simulating the flow field and particle trajectories around a two-dimensional horizontal cross section of the CIMS pylon and inlet structure (Fig. 1). The value of EF is near unity for small particles (< 0.1 μ m) and increases with particle size, as found for similar configurations [Northway et al., 2002]. For particles larger than approximately 10 μ m in diameter, typical of cirrus cloud ice particles sampled during CRYSTAL-FACE, EF for the front inlet approaches the maximum value of U_0/U . Since both CIMS channels sample at a constant mass flow of 1.85 standard liters per minute (slpm), U, and therefore, EF, are dependent upon the ambient temperature and pressure. Under typical WB-57F sampling conditions during CRYSTAL-FACE (temperature = 213 K, pressure = 170 hPa, U_0 = 200 m·s⁻¹), EF has a

maximum value of approximately 16.

2 Cirrus cloud particles entering the front CIMS inlet travel through a 20 cm length of Teflon™ tubing (6.4 mm outside diameter, 4.0 mm inside diameter) upstream of the 4 CIMS flow control valve and flow tube (Fig. 1). The use of Teflon™ sample lines ensures that HNO₃ will not readily absorb on the inlet surfaces [Neuman et al., 1999]. 6 This tubing, which is heated to 48° C in flight, has two bends to help ensure that large particles entering the inlet will impact on the tubing walls and subsequently evaporate 8 prior to reaching the flow control valve. Particles with diameters greater than approximately 20 μ m have large enough stopping distances at the freestream velocity that 10 they impact at the first bend. Some particles that do not fully evaporate will impact in the body of the flow control valve or the flow tube entrance. HNO3 condensed on the 12 particle surfaces is liberated to the gas phase early in the evaporation process and measured as a gas-phase equivalent volume mixing ratio. The HNO₃ mixing ratio 14 measured by the front CIMS channel, therefore, represents the sum of the gas- and the condensed-phase values, with the condensed-phase component enhanced by the value of 16 EF.

3. Observations

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Condensed-phase HNO₃ was observed coincident with cirrus cloud observations on 4 WB-57F science flights conducted as part of CRYSTAL-FACE. These flights originated and terminated at the United States Naval Air Facility, Key West (24.6° N, 81.7° W) in Florida on July 11, 13, 19 and 21, 2002. Time series data of HNO₃ mixing ratios observed from the front and bottom CIMS channels (HNO₃|_{front} and HNO₃|_{gas}, respectively), as well as particle SAD, IWC (represented as a gas-phase equivalent volume mixing ratio) and meteorological parameters are shown for July 11, 13, 19 and 21 in Figs. 2-5, respectively. The presence of cirrus cloud particles is indicated by increases in SAD and IWC above background values. The presence of condensed-phase HNO₃ in a flight segment is indicated by HNO₃|_{front} values that are significantly greater than

HNO₃l_{gas} values. Flight segments identified by purple bars in panel (b) for July 13 (Fig. 3) and July 19 (Fig. 4) represent the observation of contrail cirrus clouds. As stated previously, these clouds are characterized by having high particle number densities with volume-weighted mean diameters typically much lower than cirrus clouds formed by natural processes. Due to the uncertainties in the SAD measurements in the contrail-formed cirrus clouds and in the value of EF for particles in this size range, these clouds are not considered in the data analysis presented here.
Cirrus clouds were observed from the WB-57F at pressures between 122 hPa and 224

hPa during the flights of July 11, 13, 19 and 21, corresponding to pressure altitudes between 11 km and 15 km (Figs. 2-5). These clouds were observed at temperatures between 197 K and 224 K. Figures 2-5 show the strong temporal correlation of (HNO₃|_{front} - HNO₃|_{gas}) with both SAD and IWC in cirrus clouds, with HNO₃|_{gas} approaching zero during a number of cirrus events. Outside of clouds, measured values of HNO₃|_{front} and HNO₃|_{gas} generally agree well (with an overall correlation coefficient, r, of 0.92). However, some periods in Figs. 2-5 show offsets between the two channels that are best explained as changes in the inlet line surfaces during the flight. Figure 4 also indicates elevated values of relative humidity (with respect to ice) during a number of cirrus cloud encounters, as described by *Gao et al.* [2003].

Condensed-phase HNO₃, proportional to the difference between the values of HNO₃|_{front} and HNO₃|_{gas}, was observed primarily at SADs greater than 200 μ m²·cm⁻³ during CRYSTAL-FACE (Fig. 6). Note that for the values of (HNO₃|_{front} -HNO₃|_{gas}) shown in Fig. 6, HNO₃|_{front} is not corrected for particle oversampling. Values of (HNO₃|_{front} -HNO₃|_{gas}) at SADs less than 200 μ m²·cm⁻³ are near the detection limit and highly variable due to CIMS instrument noise. In the analyses presented here, observations at SADs greater than 200 μ m²·cm⁻³ are selected to represent measurements made in cirrus clouds (shown by the dashed line in Fig. 6).

3.1. Quantifying Condensed-Phase HNO₃

As stated previously, HNO₃|_{front} represents the sum of gas-phase and condensedphase HNO₃, with the condensed-phase component enhanced by the value of *EF*. The
amount of HNO₃ condensed on cirrus cloud particles (HNO₃|_{con}) can therefore be
calculated according to equation (1),

$$HNO_3|_{con} = \frac{HNO_3|_{front} - HNO_3|_{gas}}{EF}$$
 (1)

6 where HNO₃|con is reported as a gas-phase equivalent volume mixing ratio with a precision of 3 pptv (1 σ , 10-s averages). The use of equation (1) in calculating HNO₃|_{con} 8 is illustrated in Fig. 7 for a cirrus cloud encounter by the WB-57F on July 13, 2002. Increases in SAD and IWC during this cloud event are accompanied by an increase in 10 HNO₃|_{front} above the gas-phase value of approximately 0.5 ppbv (Fig. 7a,d,e). Accounting for the particle enhancement factor of approximately 13.7 using equation (1), 12 maximum values of HNO₃|_{con} during this cloud event approached 0.1 ppbv (Fig. 7b,c). It should be noted here that calculated values of HNO₃|con in cirrus clouds are not always 14 consistent with the observed decreases in HNO₃ l_{gas} that result from HNO₃ uptake. Quantitative agreement between HNO₃|con and deficits in HNO₃|gas can only occur if the 16 cloud particles are sampled in the same air mass in which uptake occurred. Due to gravitational settling, however, cirrus particles may sediment into air masses that may be 18 more or less depleted in gas-phase HNO₃ at the time of sampling.

3.2. Cirrus Cloud Particle Measurements

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Cirrus cloud particle SAD was derived from measurements provided by both the CAPS and CIN instruments onboard the WB-57F during CRYSTAL-FACE. SAD was derived from the CAPS data by integrating particle size distribution and number density measurements, while bulk measurements of cloud extinction coefficient at 635 nm were used to derive SAD from the CIN data. A comparison between the CIN- and CAPS-derived SADs indicates good agreement between the two instruments for the flights on

July 11, 19 and 21, with SADs derived from the CIN measurements ranging from 23-2 39% higher than the CAPS-derived values on those 3 flight days (Fig. 8). On July 13, the CIN measurements were 54% higher than the CAPS values, which may be attributable to 4 the sampling of optically thin subvisual cirrus clouds near the tropopause on that day. Cloud extinction in these subvisual cirrus is close to the sensitivity threshold of the CIN. 6 The observed differences are nonetheless within the combined uncertainties of the two instruments on all 4 flight days. The analyses presented here make use of SADs derived 8 from the CAPS measurements. Due to the fact that the SADs derived from the CIN measurements are 23-54% higher than the CAPS values, the SADs utilized here can be 10 considered lower limits. As stated previously, the value of EF used in calculating HNO₃|_{con} approaches a

As stated previously, the value of EF used in calculating HNO₃|_{con} approaches a maximum value of U_0/U (the ratio of the WB-57F true air speed to the air velocity in the sample inlet) at particle sizes greater than approximately 10 μ m. SADs calculated from the CAPS measurements on the 4 flight days considered here (in cirrus clouds with total SADs greater than 200 μ m²·cm⁻³) indicate that 92±9% of the surface area resides on particles larger than 10 μ m in diameter. Use of the maximum value of EF in calculating HNO₃|_{con} via equation (1), therefore, is expected to introduce no more than 10% uncertainty into the value of HNO₃|_{con}.

The VMD of cirrus cloud particles observed during CRYSTAL-FACE ranged from approximately 3 μ m up to 700 μ m, with most clouds having VMDs greater than 20 μ m (Fig. 9a). The largest particles (> 500 μ m) were observed primarily in clouds with SADs greater than $10^4 \ \mu$ m²·cm⁻³. As expected, IWC shows a strong correlation with SAD (Fig. 9b). IWCs as high as 1000 ppmv were observed during some cloud events (Fig. 9b). The highest values of VMD and IWC were observed from the WB-57F primarily at temperatures between 205 K and 215 K during the flights shown in Fig. 9.

4. Discussion

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4.1. HNO₃ Uptake on Cirrus Cloud Particles

The coincident observation of cirrus clouds and condensed-phase HNO₃ during 2 CRYSTAL-FACE is assumed here to result from the uptake of HNO₃ on the surface of cirrus cloud particles. Laboratory studies indicate that the low solubility of HNO₃ in ice 4 will not allow a significant fraction of HNO₃|con to reside in the bulk of the cirrus particles [Sommerfeld et al, 1998, Hanson and Ravishankara, 1991]. Furthermore, 6 Dominé and Thibert [1996] have suggested that the high diffusivity of HNO₃ in ice is such that HNO₃ trapped in the bulk ice during particle formation will migrate to the 8 particle surface. We note, however, that the measurements presented here cannot distinguish between surface uptake and HNO₃ that may be condensed in the bulk of the 10 particles. HNO₃ uptake on cirrus cloud particles can be represented in terms of molecular coverage, given by the ratio of HNO₃|con to SAD in units of molecules cm⁻². 12 HNO₃ surface coverages observed during CRYSTAL-FACE are shown as a function of temperature in Fig. 10, with symbols colored according to P(HNO₃) (see legend). Data 14 shown by triangles at temperatures less than 200 K in Fig. 10 represent observations under conditions in which nitric acid trihydrate (NAT) is stable, as predicted by ambient 16 temperature and the ratio of HNO₃|_{gas} to H₂O vapor [Hanson and Mauersberger, 1988, Gao et al., 2003]. It has been proposed that, under conditions in which NAT is stable, 18 HNO₃ forms NAT clusters or layers on the particle surface which interfere with the condensation of H₂O molecules on the particle surface, and thereby increase the relative 20 humidity with respect to ice in the cirrus cloud [Gao et al., 2003]. The mean HNO₃ coverage observed during CRYSTAL-FACE was 1.9·10¹³ 22 molecules·cm⁻², with maximum coverages reaching 1.4·10¹⁴ molecules·cm⁻² during a few cirrus cloud events (Fig. 10). While the greatest coverages were observed at 24 temperatures between 205 K and 210 K, mean coverages binned according to temperature show no temperature dependence above 200 K (black symbols in Fig. 10). 26 The average value for measurements between 195 K and 200 K is approximately a factor of 3 greater than values above 200 K. Generally higher HNO₃ coverages at lower

temperatures have been observed in field measurements reported by both Kondo et al. [2003] and Ziereis et al. [2003]. HNO₃ coverages show a minimal dependence on 2 $P(\text{HNO}_3)$, with the lowest coverages occurring at $P(\text{HNO}_3)$ values below 2.5·10⁻⁸ hPa 4 (Fig. 10). A number of laboratory studies have also reported increased HNO₃ coverages with increasing $P(HNO_3)$, albeit at $P(HNO_3)$ values substantially higher than those 6 presented here (5.0·10⁻⁷ - 3.0·10⁻⁶ hPa) [Hudson et al., 2002, Hynes et al., 2002]. These results highlight the need for a comprehensive laboratory study of HNO₃ uptake on ice 8 surfaces at $P(HNO_3)$ values below $2.0 \cdot 10^{-7}$ hPa that are typical of the subtropical UT. The coverage of HNO₃ on ice, in general, can be modeled or predicted using the 10 kinetics or thermodynamics of the uptake process [Gao et al., 2003, Hudson et al., 2002]. Using laboratory measurements and a semi-empirical equilibrium surface coverage 12 model, Hudson et al. [2002] have predicted HNO3 coverage on ice surfaces as a function of temperature and $P(HNO_3)$. This multilayer Frenkel-Halsey-Hill (FHH) model was 14 fitted to equilibrium HNO₃ coverages observed on vapor-deposited ice films at temperatures between 213 K and 219 K with a P(HNO₃) of 1.1·10⁻⁶ hPa. HNO₃ surface 16 coverages predicted by the FHH model are shown as a function of temperature in Fig. 11a, together with the HNO3 coverages observed in cirrus clouds during CRYSTAL-18 FACE. The isobaric lines representing the modeled coverages are colored on the same scale as the observed coverages according to the values of $P(HNO_3)$ input to the model. 20 Fig 11a indicates better agreement between the modeled and observed HNO₃ coverages at temperatures higher than approximately 205 K, while at lower temperatures the 22 modeled coverages increase to values far greater than those observed at comparable temperatures and P(HNO₃) values. The high model coverages below 205 K may result 24 from the fact that the model was fitted to laboratory data at temperatures above 213 K, and the coverages presented here, therefore, are extrapolated to lower temperatures where 26 the uncertainty in the model increases.

A number of studies have described the uptake of HNO₃ on ice surfaces using a

Langmuir surface chemistry model [Tabazdeh et al., 1999, Hynes et al., 2002, Meier et al., 2002]. The Langmuir isotherm predicts the fractional HNO₃ surface coverage (θ) according to equation 2,

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$$\theta = \frac{K_{eq}^{1/2} \cdot P(\text{HNO}_3)^{1/2}}{1 + K_{eq}^{1/2} \cdot P(\text{HNO}_3)^{1/2}}$$
 (2)

where K_{eq} represents the equilibrium adsorption constant, given by the ratio of the rates 6 of adsorption and desorption (k_a/k_d) [Laidler and Meiser, 1982]. The value of θ is unity when the HNO₃ surface coverage reaches a complete monolayer (1.0·10¹⁵ molecules·cm⁻ 8 2). We note that the surface density of HNO₃ molecules when forming a complete monolayer is somewhat uncertain, and the density of 1.0·10¹⁵ molecules·cm⁻² stated here 10 should be considered an upper limit [Hudson et al., 2002]. If the HNO3 surface density is lower than 1.0·10¹⁵ molecules·cm⁻² for a complete monolayer, the resulting fractional 12 surface coverages will be higher than those stated here. Application of the dissociative form of the Langmuir isotherm is supported here by spectroscopic studies of HNO₃ 14 uptake on thin ice films at 211 K, which indicate, by the presence of H₃O⁺ and NO₃⁻ ions on the ice surface, that HNO₃ dissociates upon adsorption [Zondlo et al., 1997]. The 16 temperature-dependent equilibrium adsorption constant in equation (2) can be calculated according to equation (3) [Adamson and Gast, 1997],

$$K_{eq} = \frac{100 \cdot N_{A} \cdot \sigma_{0} \cdot \tau_{0}}{(2\pi \cdot M \cdot R \cdot T)^{1/2}} e^{\left(\frac{-\Delta H_{ads}}{c \cdot R \cdot T}\right)} \text{ hPa}^{-1}$$
 (3)

where N_A is Avogadro's number $(6.02 \cdot 10^{23} \text{ mol}^{-1})$, σ_0 is the area of one adsorption site (10^{-19} m^2) , τ_0 is the time constant for adsorbate oscillation perpendicular to the surface (10^{-13} s) , M is the molecular weight of HNO₃ $(0.063 \text{ kg} \cdot \text{mol}^{-1})$, R is the ideal gas constant $(8.314 \text{ J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1})$, T is temperature (in K), c is a unit conversion factor $(2.39 \cdot 10^{-4} \text{ kcal} \cdot \text{J}^{-1})$ and ΔH_{ads} is the adsorption enthalpy of HNO₃ on ice (in kcal·mol⁻¹).

Using equations (2) and (3), a Langmuir isotherm was fitted to the CRYSTAL-FACE 2 observations of fractional surface coverage and P(HNO₃) (Fig. 12). Using the median temperature of 208 K for the observations shown in Fig. 9, the best fit to the experimental 4 data was achieved with a ΔH_{ads} of -11.0 kcal·mol⁻¹, or 46.0 kJ·mol⁻¹ (red line in Fig. 12). Note that the data are largely bound by ΔH_{ads} values of -10.0 kcal·mol⁻¹ and -12.06 kcal·mol⁻¹. Also shown in Fig. 12 are Langmuir isotherms at the same temperature for ΔH_{ads} values of -14.2 kcal·mol⁻¹ and -12.9 kcal·mol⁻¹ reported by *Tabazadeh et al.* 8 [1998] and Hynes et al., [2002], respectively. Fractional coverages predicted using these previously published values of ΔH_{ads} far exceed the coverages observed during 10 CRYSTAL-FACE, indicating these adsorption enthalpies are too high to accurately describe the observations presented here. Bartels-Rausch et al. [2002], using a 12 chromatographic technique, have recently reported a ΔH_{ads} for HNO₃ uptake on ice of -10.5 kcal·mol⁻¹ that is in good agreement with the value of -11.0 kcal·mol⁻¹ presented 14 here. We caution that the effective ΔH_{ads} reported here is empirically derived from observations in a dynamic system which may or may not be in steady state, and this 16 value, therefore, cannot be considered a fundamental thermodynamic parameter. Nonetheless, the Langmuir formalism, using a ΔH_{ads} of -11.0 kcal·mol⁻¹, effectively 18 describes the CRYSTAL-FACE observations of HNO₃ uptake on cirrus cloud particles to within a factor of 5 (Fig. 12). It should be noted that the data shown in Fig. 12 span a 20 temperature range form 197 K to 218 K, and that the Langmuir isotherms were fitted at the median temperature of 208 K. Use of a single temperature in fitting the isotherms is 22 supported by the results of Hynes et al. [2002], who reported a variation of less than 2% between values of ΔH_{ads} derived from laboratory measurements at 218 K and 228 K. 24 Having derived an effective value of ΔH_{ads} for HNO₃ adsorption on cirrus cloud particles in the UT, HNO₃ coverages predicted by the Langmuir surface chemistry model 26 (as a function of temperature) can be compared to the CRYSTAL-FACE observations (Fig. 11b). As in Fig. 11a, the isobaric lines are colored on the same scale as the

observed coverages. The calculated coverages shown in Fig. 11b indicate that the model does not adequately describe the considerable variability in the observed coverages at a given temperature and $P(\text{HNO}_3)$. Nonetheless, when using the empirically derived ΔH_{ads} of -11.0 kcal·mol⁻¹, the Langmuir model is capable of predicting the observed

coverages within a factor of 5 or better. The variability in the observed coverages, and

the less than perfect agreement with the uptake models, can be explained, in part, if the adsorbed HNO₃ is not in equilibrium with HNO₃ in the gas phase. Previous field studies

have also shown HNO₃ surface coverages to be highly variable throughout the temperature and *P*(HNO₃) ranges observed [*Kondo et al.*, 2002, *Ziereis et al.*, 2003].

4.2. HNO₃ Partitioning in Cirrus Clouds

The fraction of total HNO₃ present on cirrus cloud particles was observed to increase with SAD during CRYSTAL-FACE (Fig. 13). The mean value of HNO₃ partitioned in the condensed phase at SADs greater than 200 μ m²·cm⁻³ was 16%. Up to 100% of the total HNO₃ was partitioned on ice particles during some cirrus cloud encounters, at SADs between 350 and 4.2·10⁴ μ m²·cm⁻³ and temperatures between 201 K and 213 K. Measurements reported by *Ziereis et al.* [2003] in midlatitude cirrus clouds reveal a similar relationship between condensed-phase NO_y partitioning and SAD, although maximum reported values of condensed-phase NO_y partitioned in cirrus clouds did not exceed 50% of the total observed NO_y. Measurements of NO_y uptake in a mountain wave cirrus cloud reported by *Weinheimer et al.* [1998] indicate complete uptake of HNO₃, provided that the ambient HNO₃/NO_y ratio in the cloud was 0.1-0.2. Gas-phase HNO₃ was not measured in either of these previous studies, making an accurate assessment of the fraction of HNO₃ remaining in the gas phase after uptake difficult.

Krämer et al. [2003] have recently studied the partitioning of HNO₃ in Arctic cirrus clouds, and have modeled the role of HNO₃ uptake by interstitial HNO₃-H₂SO₄-H₂O ternary solution aerosols in partitioning. This study concluded that some fraction of the total HNO₃ in Arctic cirrus clouds must remain in the gas phase, with the remainder

partitioned predominately in interstitial aerosols at temperatures less than 205 K when SADs are low, and on cirrus cloud particles at higher SADs. Measurements in subtropical cirrus clouds reported here, however, indicate that up to 100% of the total

4 HNO₃ can be partitioned in cirrus ice particles both at low temperatures and low SADs. Furthermore, we see no evidence of significant uptake of HNO₃ in ternary solution

aerosols outside of clouds in the subtropical UT during the flights considered in this study. There is evidence, however, of HNO₃ uptake by ternary solution aerosols in the near absence of cirrus ice particles on at least one other CRYSTAL-FACE flight (July 9,

2002) [Weinheimer et al., 2002]. We also note that HNO₃ may be contained in a ternary solution on the surface of the cirrus ice particles.

4.3. HNO₃ and HNO₃/NO_y in the Cloud-Free Upper Troposphere and Lower Stratosphere

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Measurements in the cloud-free subtropical UT during CRYSTAL-FACE indicate that the gas-phase HNO₃/NO_y ratio is highly variable, ranging from zero to approximately 0.5 (colored symbols in Fig. 14a). The HNO₃/NO_y ratio is generally higher and also variable in the subtropical lower stratosphere (LS), with values observed between 0.05 and 1 (black symbols in Fig. 14a). HNO₃ is expected to be the predominate NO_y species in the LS away from the tropopause region [*Neuman et al.*, 2001]. The lower HNO₃/NO_y ratios (< 0.3) observed in the UT are affected by low observed values of HNO₃, due to HNO₃ removal by uptake and sedimentation by cloud particles in cirrus processed air masses (yellow symbols in Fig. 14b), or from elevated levels of NO_y due to NO production from lightning strikes (purple symbols in Fig. 14b). Previously measured values of the HNO₃/NO_y ratio in the midlatitude UT over the continental United States were also highly variable and ranged from approximately 0.1 to 0.5 [*Neuman et al.*, 2001]. The large range and variability of HNO₃/NO_y ratios observed in the UT during CRYSTAL-FACE highlights the value in measuring gas-phase HNO₃ when assessing HNO₃ uptake on cirrus cloud particles, over deriving gas-phase HNO₃ from measured

NO_y and a constant assumed HNO₃/NO_y ratio.

A number of cirrus clouds observed in the subtropical UT during CRYSTAL-FACE had VMDs between 200 μm and 700 μm. Terminal fall velocities for cirrus ice particles in this size range are 1 m·s·1 to 10 m·s·1 (Fig. 9a) [Meier and Hendricks, 2002]. With 16% of the total HNO3 in cirrus clouds adsorbed on ice particles, the gravitational redistribution of a significant fraction of the total HNO3 in these clouds can therefore occur on a timescale of minutes to hours. Cirrus clouds with up to 100% of the total HNO3 partitioned in the condensed phase have even greater potential to redistribute HNO3 in the UT.

5. Conclusions and Implications

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A number of cirrus cloud encounters in the UT by the NASA WB-57F during CRYSTAL-FACE were accompanied by the observation of condensed-phase HNO₃. Maximum levels of condensed-phase HNO₃ exceeded the gas-phase equivalent of 0.8 ppbv during some cirrus events. A mean HNO₃ surface coverage of 1.9·10¹³ molecules·cm⁻² was observed on the flights of July 11, 13, 19 and 21, 2002, with maximum surface coverages reaching as high as 1.4·10¹⁴ molecules·cm⁻² during a few cirrus cloud encounters. Molecular coverages predicted using a Langmuir surface chemistry model agree with the observed coverages to within a factor of 5 or better when using an empirically derived ΔH_{ads} of -11.0 kcal·mol⁻¹. The mean percentage of total HNO₃ condensed on cirrus cloud particles was 16%, with up to 100% of the HNO₃ partitioned in the condensed phase in a number of cirrus clouds. The fraction of total HNO₃ in the condensed phase was found to increase strongly with SAD. Based on the large diameters of cloud particles containing HNO₃ observed during CRYSTAL-FACE, the redistribution of HNO₃ in the UT will be very effective in some cloud systems. The interpretation of future observations of HNO₃ uptake on cirrus particles will be improved by a knowledge of the individual history of the air parcels in which the cirrus clouds are formed [Kärcher, 2003] and laboratory studies of HNO3 uptake at low gas-phase

abundances ($P(HNO_3) < 2.0 \cdot 10^{-7} \text{ hPa}$) and low temperatures (195 K – 220 K).

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	Figure Captions
2	
4	Figure 1. Schematic diagram of the NOAA CIMS inlet pylon. Labeled components are as follows: (a) front channel inlet, (b) bottom channel inlet, (c) zero gas addition, (d)
6	calibration gas addition, (e) reagent gas carrier addition, (f) reagent gas addition, (g) HNO ₃ permeation cell (calibration standard), (h) flow control valve motor, (i) flow control valve body, (j) flow tube, (k) ion source, (l) flow straightener. For clarity,
8	complete components are shown only for the front channel. Inlet lines are constructed of Teflon™ tubing (6.4 mm outside diameter, 4.0 mm inside diameter), heated to
10	temperatures of 40° C (bottom channel inlet) and 48° C (front channel inlet) to avoid wall losses. The higher temperature of the front inlet ensures that ice particles entering the
12	inlet evaporate upon impaction with the tubing wall. The operating principles of the NOAA CIMS instrument have been described in detail by <i>Neuman et al.</i> [2000].
14	
16	Figure 2. Time series measurements of the HNO ₃ mixing ratio observed from the front and bottom CIMS channels (HNO ₃ _{front} and HNO ₃ _{gas} , respectively,) on the flight of July 11, 2002. Note that values of HNO ₃ _{front} do not include a correction for particle
18	oversampling in the front channel inlet. Discontinuities in the time series result from CIMS instrument calibrations and other housekeeping procedures. Also shown are SAD,
20	IWC, water vapor, relative humidity (with respect to ice), and ambient temperature and pressure. All data are represented as 10-s averages. Minor divisions on the horizontal
22	scale represent 15 minutes (or approximately 150 km) of flight.
24	Figure 3. Same as Fig. 2, for the flight on July 13, 2002. Purple bars in panel (b) represent 3 flight segments in which cirrus clouds formed in the contrail of the WB-57F
26	were observed. The inset in panel (a) shows 1-s averages of HNO ₃ _{front} and HNO ₃ _{gas} during the first contrail intercept. The purple bar in the inset represents the same time
28	period as the first purple bar in panel (b). The vertical scale on the inset panel is 0 to 1 ppbv. Measurements of HNO ₃ _{front} and HNO ₃ _{gas} are incomplete during the second and
30	third contrail intercepts due to instrument housekeeping procedures.
32	Figure 4. Same as Fig. 2, for the flight on July 19, 2002. Purple bars in panel (b) represent 2 flight segments in which cirrus clouds formed in the contrail of the WB-57F
34	were observed. The insets in panel (a) shows 1-s averages of HNO ₃ _{front} and HNO ₃ _{gas} during the two contrail intercepts. The purple bars in the first and second insets represent
36	the same time periods as the first and second purple bars in panel (b). The vertical scale on the inset panels is 0 to 0.5 ppbv.
38	
4.0	Figure 5. Same as Fig. 2, for the flight on July 21, 2002.
40	Figure 6 (HNOsic HNOsic HNOsic) vs. SAD for the flights of July 11, 13, 10 and 21
42	Figure 6. (HNO ₃ _{front} - HNO ₃ _{gas}) vs. SAD for the flights of July 11, 13, 19 and 21, 2002. Values of (HNO ₃ _{front} -HNO ₃ _{gas}) are proportional, but not equal to condensed-phase HNO ₃ (HNO ₃ _{con}) because they do not include a correction for particle

oversampling in the front channel inlet. Red circles represent 10-s averages and black squares represent mean values of the 10-s data grouped into deciles. Vertical bars represent the standard deviation about the mean value in each decile, and horizontal bars indicate the upper and lower boundaries of each decile. Note that the vertical axis is shown with a logarithmic scale at values greater than 2.5. The dashed vertical line at the SAD of 200 μ m²·cm⁻³ represents the lower limit chosen here to represent measurements made in cirrus clouds.

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Figure 7. Calculation of HNO₃|_{con} for a cirrus cloud encounter by the NASA WB-57F on July 13, 2002, showing HNO₃|_{front} and HNO₃|_{gas} (panel a), EF (panel b), HNO₃|_{con} (panel c), SAD (panel d) and IWC (panel e). All data are shown as 10-s averages. The horizontal axis spans approximately 16 minutes and 180 km of flight.

Figure 8. Comparison of CIN- and CAPS-derived SAD measured during the flights of July 11, 13, 19 and 21, 2002. Data at SADs less than 3 μm²·cm⁻³ are expanded in the inset. All data are shown as 10-s averages. Lines represent a least-squares fit to the data

for each flight date. The slope of the linear fit (constrained through the origin) is shown

for each flight date in the figure legend.

Figure 9. Panel (a). Volume-weighted mean diameter of cirrus cloud particles vs. SAD measured during the flights on July 11, 13, 19 and 21, 2002. Only data at SADs greater

than 200 μm²·cm⁻³ are shown. Data are colored according to ambient temperature. The black line represents the terminal fall velocity for ice particles in the upper troposphere, calculated according to *Meier and Hendricks* [2002]. Panel (b). IWC (represented as a

calculated according to *Meier and Hendricks* [2002]. Panel (b). IWC (represented as a gas-phase equivalent volume mixing ratio) vs. SAD. At 130 hPa and 200 K, an IWC of

100 ppmv is equivalent to an ice water concentration of 14 mg·m⁻³. Other details same as panel (a).

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- Figure 10. HNO₃ coverage vs. temperature for measurements made in cirrus clouds at SADs greater than 200 μ m²·cm⁻³ on July 11, 13, 19 and 21, 2002. Symbols are colored according to $P(\text{HNO}_3)$. Triangles at temperatures less than 200 K indicate measurements
- made under conditions in which NAT is stable. Mean values of HNO₃ coverage are plotted at the mean of 5 K temperature bins from 195 K to 220 K (black squares). Error

bars represent the standard deviation in each temperature bin. Negative values of HNO₃ coverage are included in the calculation of the mean values. The dashed line at 1.9·10¹³

molecules·cm⁻² represents the mean HNO₃ coverage observed during CRYSTAL-FACE. A complete HNO₃ monolayer is formed when the coverage reaches 1.0·10¹⁵

38 molecules·cm⁻².

- Figure 11. HNO₃ coverage vs. temperature for measurements made in cirrus clouds at SADs greater than $200 \ \mu \text{m}^2 \cdot \text{cm}^{-3}$ on July 11, 13, 19 and 21, 2002. Symbols are colored
- 42 according to $P(HNO_3)$. Triangles at temperatures less than 200 K indicate measurements made under conditions in which NAT is stable. Data are represented as 10-s averages.

Panel (a). Lines are isobars representing HNO₃ coverages calculated by the FHH HNO₃ 2 uptake model [Hudson et al., 2002] and are colored according to the same temperature scale as the observed coverages (see legend). Panel (b). Same as panel (a), except the 4 isobars represent HNO₃ coverages calculated by the Langmuir surface chemistry model. 6 Figure 12. Fractional HNO₃ surface coverage (θ) vs. $P(HNO_3)$ measured during the flights of July 11, 13, 19 and 21, 2002. Colored lines are isotherms fitted according to 8 equations (2) and (3) (see text) with values of ΔH_{ads} shown in the legend and the median observed temperature of 208 K. All data are represented as 10-s averages for 10 measurements made at SADs greater than 200 µm²·cm⁻³. 12 Figure 13. Fraction of total HNO₃ in the condensed phase vs. SAD measured during the flights of July 11, 13, 19 and 21, 2002. Data are represented as 10-s averages for measurements made at SADs greater than 200 μ m²·cm⁻³ and are colored according to 14 temperature. Black squares represent mean values of the 10-s data grouped into quintiles. 16 Vertical bars represent the standard deviation about the mean value in each quintile and horizontal bars represent the upper and lower boundaries of each quintile. Values of 18 condensed-phase partitioning greater than 100% occur when zero or near-zero HNO₃ l_{pas} abundances are measured as negative values. The dashed line at 16% represents the 20 mean value of HNO₃ partitioned in the condensed phase during CRYSTAL-FACE. 22 Figure 14. Panel (a). Vertical profile of the gas-phase HNO₃/NO_y ratio observed in cloud free air (SADs less than 20 μ m²·cm⁻³) during the flights of July 11, 13, 19 and 21, 24 2002. Tropospheric measurements (according to the Microwave Temperature Profiler) are colored according to measured NO. Stratospheric measurements are shown in black. 26 Panel (b). Vertical profile of HNO₃|_{gas}. Other details same as panel (a).

Table 1. Measurement details.

Measurement	Institution	Precision ^a	Accuracy	Reference.
HNO ₃	NOAA Aeronomy Lab	30 pptv, 10 pptvb	±20%	Neuman et al., 2002
SAD - CAPS	Univ. Nacional Autonoma de Mexico	-	±50%	Baumgardner et al., 2002
SAD - CIN	University of Utah	-	±15%	Gerber et al., 2000
IWC	Harvard University	0.7 ppmv	±17%	Weinstock et al., 2004
Water Vapor	Harvard University	0.5 ppmv	±5%	Weinstock et al., 1994
NO	NCAR	5 pptv	±6%	Weinheimer et al., 2004
NO _y	NCAR	8 pptv	±12%	Weinheimer et al., 2004
Temperature	NASA Ames	0.01 K	±0.3 K	Scott et al., 1990
Pressure	NASA Ames	0.1 hPa	±0.3 hPa	Scott et al., 1990
True Air Speed	NASA Ames	0.1 m·s ⁻¹	±1 m·s-1	Scott et al., 1990

a Precision values are reported for 10-s averages

b 10 pptv is the precision for condensed-phase HNO₃.

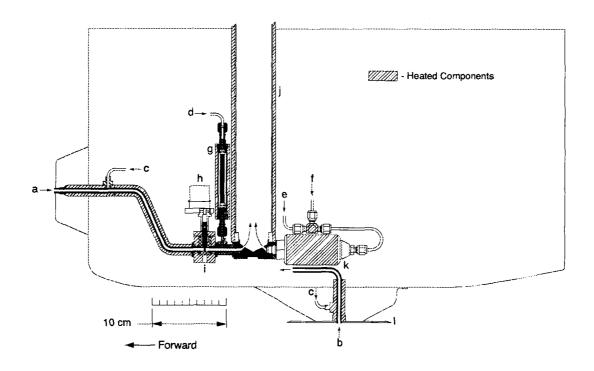


Fig. 1 Popp et al.

Figure 1. Schematic diagram of the NOAA CIMS inlet pylon. Labeled components are as follows: (a) front channel inlet, (b) bottom channel inlet, (c) zero gas addition, (d) calibration gas addition, (e) reagent gas carrier addition, (f) reagent gas addition, (g) HNO₃ permeation cell (calibration standard), (h) flow control valve motor, (i) flow control valve body, (j) flow tube, (k) ion source, (l) flow straightener. For clarity, complete components are shown only for the front channel. Inlet lines are constructed of Teflon[™] tubing (6.4 mm outside diameter, 4.0 mm inside diameter), heated to temperatures of 40° C (bottom channel inlet) and 48° C (front channel inlet) to avoid wall losses. The higher temperature of the front inlet ensures that ice particles entering the inlet evaporate upon impaction with the tubing wall. The operating principles of the NOAA CIMS instrument have been described in detail by *Neuman et al.* [2000].

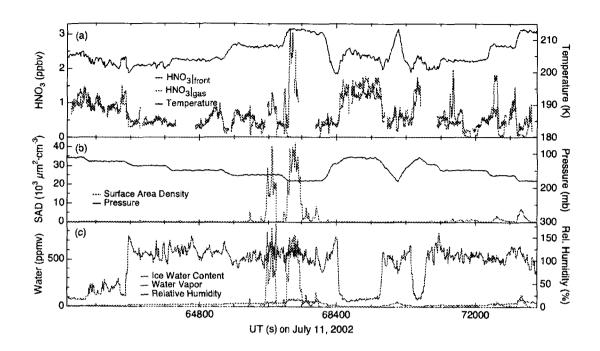


Fig. 2 Popp et al.

Figure 2. Time series measurements of the HNO₃ mixing ratio observed from the front and bottom CIMS channels (HNO₃|_{front} and HNO₃|_{gas}, respectively,) on the flight of July 11, 2002. Note that values of HNO₃|_{front} do not include a correction for particle oversampling in the front channel inlet. Discontinuities in the time series result from CIMS instrument calibrations and other housekeeping procedures. Also shown are SAD, IWC, water vapor, relative humidity (with respect to ice), and ambient temperature and pressure. All data are represented as 10-s averages. Minor divisions on the horizontal scale represent 15 minutes (or approximately 150 km) of flight.

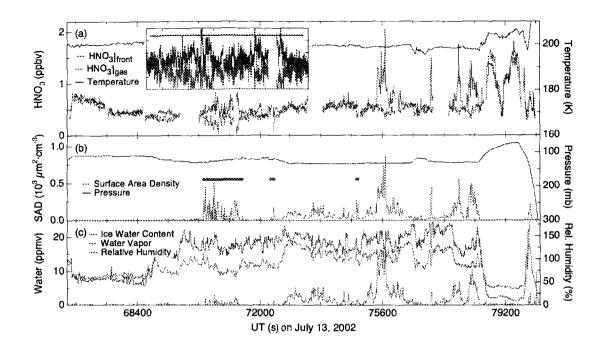


Fig. 3 Popp et al.

Figure 3. Same as Fig. 2, for the flight on July 13, 2002. Purple bars in panel (b) represent 3 flight segments in which cirrus clouds formed in the contrail of the WB-57F were observed. The inset in panel (a) shows 1-s averages of HNO₃|_{front} and HNO₃|_{gas} during the first contrail intercept. The purple bar in the inset represents the same time period as the first purple bar in panel (b). The vertical scale on the inset panel is 0 to 1 ppbv. Measurements of HNO₃|_{front} and HNO₃|_{gas} are incomplete during the second and third contrail intercepts due to instrument housekeeping procedures.

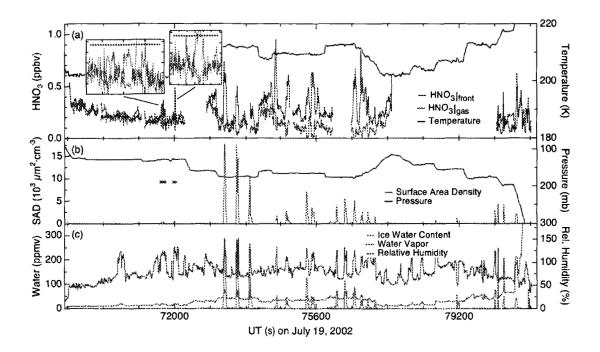


Fig. 4 Popp et al.

Figure 4. Same as Fig. 2, for the flight on July 19, 2002. Purple bars in panel (b) represent 2 flight segments in which cirrus clouds formed in the contrail of the WB-57F were observed. The insets in panel (a) shows 1-s averages of HNO₃|_{front} and HNO₃|_{gas} during the two contrail intercepts. The purple bars in the first and second insets represent the same time periods as the first and second purple bars in panel (b). The vertical scale on the inset panels is 0 to 0.5 ppbv.

10 (a) 210 Temperature (K) HNO₃ (ppbv) --- HNO_{3|front} --- HNO_{3|gas} --- Temperature Pressure (mb) 0 40 Water (10³ ppmv) SAD (10³ µm²·cm⁻³) (b) 30 20 -- Surface Area Density -- Pressure Rel. Humidity (%) 150 100 50 0 ice Water Content Water Vapor Relative Humidity 0.5 68400 72000 75600 79200 UT (s) on July 21, 2002

Fig. 5 Popp et al.

Figure 5. Same as Fig. 2, for the flight on July 21, 2002.

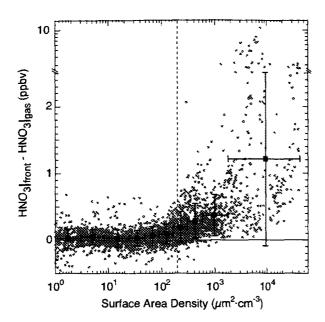


Fig. 6 Popp et al.

Figure 6. (HNO₃|_{front} - HNO₃|_{gas}) vs. SAD for the flights of July 11, 13, 19 and 21, 2002. Values of (HNO₃|_{front} -HNO₃|_{gas}) are proportional, but not equal to condensed-phase HNO₃ (HNO₃|_{con}) because they do not include a correction for particle oversampling in the front channel inlet. Red circles represent 10-s averages and black squares represent mean values of the 10-s data grouped into deciles. Vertical bars represent the standard deviation about the mean value in each decile, and horizontal bars indicate the upper and lower boundaries of each decile. Note that the vertical axis is shown with a logarithmic scale at values greater than 2.5. The dashed vertical line at the SAD of 200 μ m²·cm⁻³ represents the lower limit chosen here to represent measurements made in cirrus clouds.

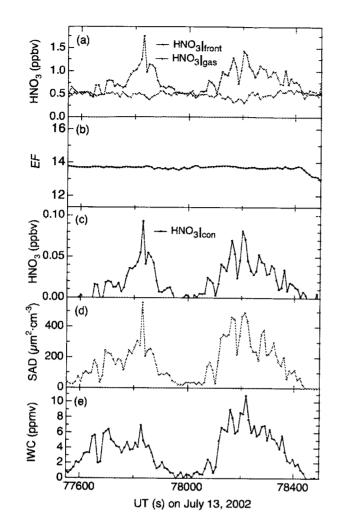


Fig. 7 Popp et al.

Figure 7. Calculation of $HNO_3|_{con}$ for a cirrus cloud encounter by the NASA WB-57F on July 13, 2002, showing $HNO_3|_{front}$ and $HNO_3|_{gas}$ (panel a), EF (panel b), $HNO_3|_{con}$ (panel c), SAD (panel d) and IWC (panel e). All data are shown as 10-s averages. The horizontal axis spans approximately 16 minutes and 180 km of flight.

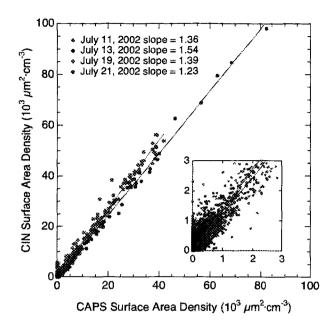


Fig. 8 Popp et al.

Figure 8. Comparison of CIN- and CAPS-derived SAD measured during the flights of July 11, 13, 19 and 21, 2002. Data at SADs less than 3 μ m²·cm⁻³ are expanded in the inset. All data are shown as 10-s averages. Lines represent a least-squares fit to the data for each flight date. The slope of the linear fit (constrained through the origin) is shown for each flight date in the figure legend.

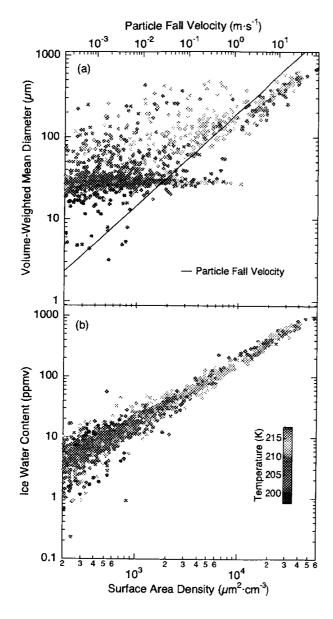


Fig. 9 Popp et al.

Figure 9. Panel (a). Volume-weighted mean diameter of cirrus cloud particles vs. SAD measured during the flights on July 11, 13, 19 and 21, 2002. Only data at SADs greater than $200 \, \mu \text{m}^2 \cdot \text{cm}^{-3}$ are shown. Data are colored according to ambient temperature. The black line represents the terminal fall velocity for ice particles in the upper troposphere, calculated according to *Meier and Hendricks* [2002]. Panel (b). IWC (represented as a gas-phase equivalent volume mixing ratio) vs. SAD. At 130 hPa and 200 K, an IWC of 100 ppmv is equivalent to an ice water concentration of 14 mg·m⁻³. Other details same as panel (a).

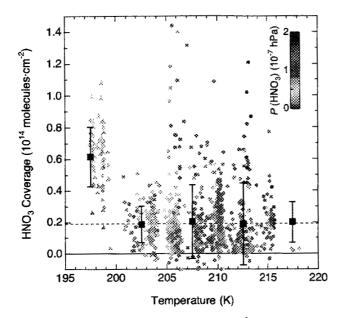


Fig. 10 Popp et al.

Figure 10. HNO₃ coverage vs. temperature for measurements made in cirrus clouds at SADs greater than 200 μm²·cm⁻³ on July 11, 13, 19 and 21, 2002. Symbols are colored according to *P*(HNO₃). Triangles at temperatures less than 200 K indicate measurements made under conditions in which NAT is stable. Mean values of HNO₃ coverage are plotted at the mean of 5 K temperature bins from 195 K to 220 K (black squares). Error bars represent the standard deviation in each temperature bin. Negative values of HNO₃ coverage are included in the calculation of the mean values. The dashed line at 1.9·10¹³ molecules·cm⁻² represents the mean HNO₃ coverage observed during CRYSTAL-FACE. A complete HNO₃ monolayer is formed when the coverage reaches 1.0·10¹⁵ molecules·cm⁻².

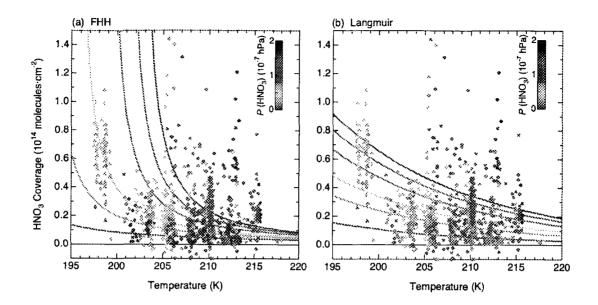


Fig. 11 Popp et al.

Figure 11. HNO₃ coverage vs. temperature for measurements made in cirrus clouds at SADs greater than 200 μ m²·cm⁻³ on July 11, 13, 19 and 21, 2002. Symbols are colored according to $P(\text{HNO}_3)$. Triangles at temperatures less than 200 K indicate measurements made under conditions in which NAT is stable. Data are represented as 10-s averages. Panel (a). Lines are isobars representing HNO₃ coverages calculated by the FHH HNO₃ uptake model [*Hudson et al.*, 2002] and are colored according to the same temperature scale as the observed coverages (see legend). Panel (b). Same as panel (a), except the isobars represent HNO₃ coverages calculated by the Langmuir surface chemistry model.

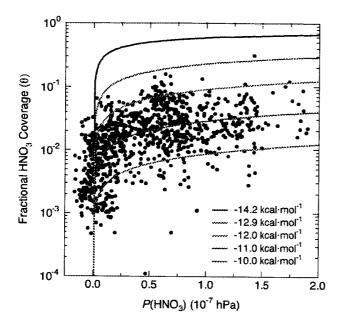


Fig. 12 Popp et al.

Figure 12. Fractional HNO₃ surface coverage (θ) vs. $P(\text{HNO}_3)$ measured during the flights of July 11, 13, 19 and 21, 2002. Colored lines are isotherms fitted according to equations (2) and (3) (see text) with values of ΔH_{ads} shown in the legend and the median observed temperature of 208 K. All data are represented as 10-s averages for measurements made at SADs greater than 200 μm^2 -cm⁻³.

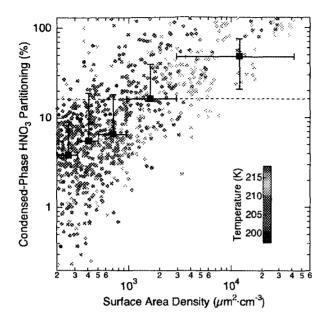


Fig. 13 Popp et al.

Figure 13. Fraction of total HNO₃ in the condensed phase vs. SAD measured during the flights of July 11, 13, 19 and 21, 2002. Data are represented as 10-s averages for measurements made at SADs greater than 200 μ m²·cm⁻³ and are colored according to temperature. Black squares represent mean values of the 10-s data grouped into quintiles. Vertical bars represent the standard deviation about the mean value in each quintile and horizontal bars represent the upper and lower boundaries of each quintile. Values of condensed-phase partitioning greater than 100% occur when zero or near-zero HNO₃l_{gas} abundances are measured as negative values. The dashed line at 16% represents the mean value of HNO₃ partitioned in the condensed phase during CRYSTAL-FACE.

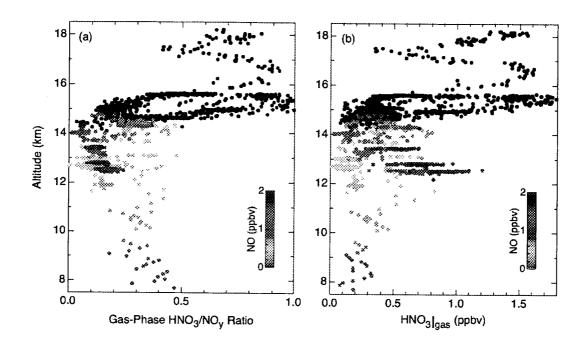


Fig. 14 Popp et al.

Figure 14. Panel (a). Vertical profile of the gas-phase HNO_3/NO_y ratio observed in cloud free air (SADs less than $20~\mu m^2 \cdot cm^{-3}$) during the flights of July 11, 13, 19 and 21, 2002. Tropospheric measurements (according to the Microwave Temperature Profiler) are colored according to measured NO. Stratospheric measurements are shown in black. Panel (b). Vertical profile of $HNO_3|_{gas}$. Other details same as panel (a).